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T182 Chemistry: Matter and Change Name Date Class Name Date Class CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY Covalent Bonding Section 9.2 Naming Molecules In your textbook, read about how binary compounds and acids are named from their formulas.

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3 17 Section 3.4 Elements and Compounds In your textbook, read about elements and compounds. Circle the letter of the choice that best completes the statement or answers the question. 1. A substance that cannot be separated into simpler substances by physical or chemical

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have similar chemical properties? 17. How is the energy level of an element's valence electrons related to its period on the periodic table? Give an example. 04' Study Guide for Content Mastery Chemistry: Matter and Change Chapter 6

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Study Guide Content Mastery Chemical
STUDY GUIDE FOR CONTENT MASTERY Name CHAPTER 10.1 continued Class STUDY GUIDE FOR CONTENT MASTERY Chemical Reactions Section 10.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence or a chemical reaction.

Matter—Properties and Changes
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Chapter 1: Introduction to Chemistry
Chemistry 7th Grade - study guide. STUDY. PLAY. Matter. Anything that has mass and takes up space. Element. A substance that cannot be separated or broken down into simpler substances by chemical means. Compound. A substance made up of atoms of two or more different elements joined by chemical bonds.

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Name CHAPTER Date Class STUDY GUIDE FOR CONTENT MASTERY ions noble gases valence Ionic Compounds Section 8.1 Forming Chemical Bonds In your textbook, read about chemical bonds and formation of ions.

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STUDY GUIDE For each of the following chemical reactions, write a word equation, a skeleton equation, and a balanced chemical equation. Be sure to show the state of each reactant and prod- uct. If you need more help writing formulas or determining the state of a substance, refer to Chapters 7 and 8 and the periodic table on pages 178—179. 17.

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STUDY GUIDE FOR CONTENT MASTERY Reaction Rates Section 17.1 A Model for Reaction Rates In your textbook, read about expressing reaction rates and explaining reactions and their rates. Use each of the terms below just once to complete the passage. coffiston-theer-ye activation-energy According to the (1) reaction rate transitiorrstate

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Chapter 18 Study Guide. STUDY. PLAY. Matter. Anything that has mass and takes up space. Compound. A pure substance made of two or more elements chemically combined. ... Number below chemical symbol or the higher of the two numbers shown. Find Atomic Number. Number in top corner of element. Chemical bonds. Holds atoms together.

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